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PHYSICO-CHEMICAL AND GEOCHEMICAL PROPER-TIES OF SOILS UNDER SAWAH SYSTEM OF INLAND VALLEYS IN NIGERIA

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ABSTRACT

This study investigated the physico-chemical and geochemical properties of soils under *sawah* in Nigeria. It was found that soils under *sawah* were majorly sandy loam to sandy clay loam having acidic reactions, low exchangeable Ca, Mg, K and Na. These soils were deficient in available P, SiO₂, S, Total Nitrogen and Total Carbon while SiO₂, Al₂O₃ and Fe₂O₃ dominated total elemental composition, accounting for a cumulative average of 96.16%. Except total elemental TiO₂ and K₂O which showed average values >1%, MnO, MgO, CaO, Na₂O and P₂O₅ showed average values <1%. Soils under *sawah* exhibited intermediated to extreme weathering degree with majority of the soil sampled falling into the category of extreme weathering. With extreme degree of weathering, rapid loss of mobile species such as basic cations from soil is imminent which may account for the results observed in this study. Thus, combination of conservative agricultural practices is recommended.

Keywords: Physico-chemical properties, geological fertilization, sawah, inland valleys

INTRODUCTION

Rice is widely grown in Nigeria under the upland rain fed, inland shallow swamps, deep water and lowland irrigated production systems (Olayemi, 1997; Oladele and Wakatsuki, 2010). However, production under these systems has not been able to meet the demand for rice in Nigeria. An average Nigerian consumes 24.8 kg of rice per year, representing 9 per cent of annual calorie intake (IRRI, 2001). There has been rapid growth in per capita rice consumption in Nigeria during the last three decades,

from 5 kg in the 1960s to 25 kg in the late 1990s (WARDA, 2003). In 1990, Nigeria imported 224,000 metric tons of rice valued at US 60 million dollars. This increased to 345,000 metric tons in 1996 with a value of US130 million dollars. By 2001, rice import increased to 1.51 million metric tons valued at US288.1 million dollars (FAO, 1994). In 2004, about 5 million tons was estimated as the total domestic rice demand while the annual domestic output of rice still hovered around 3.0 million tons, leaving about 2 million tons gap annually (NAMIS, 2004). In

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2010, Nigeria imported 2.0 million tons of rice mainly from Thailand and 73,000 tons of US parboiled rice, the highest level in the last 30 years (GAIN, 2011). At present, one of the important political-economic goals of the Nigerian government is self-sufficiency in rice production (Bello, 2004).

To improve rice production, attention must be shifted to the Inland Valleys (IVs) available across Nigeria, with high potential for lowland rice production. The values of IVs in crop production have been emphasised, especially for rice and rice-based cropping systems in West Africa (Annan-Afful et al., 2004). The IVs offer considerable potential for agricultural intensification and diversification due to their natural fertility and water availability. Efficient management and sustainable utilization of these inland valleys could, therefore, result in an increase in rice production and reduction in importation. Study has shown that only 15% or less of the total inland valleys in Nigeria has to date been under cultivation despite the agricultural potential of the inland valleys (IITA, 1990; WARDA, 1997; Abe et al., 2007) due to lack of understanding of inland valley ecosystems. Wakatsuki and Masunaga (2005) also reported that the inability to develop the lowland valleys for agriculture in West Africa accounted for the failure in achieving Green Revolution (GR). To this end, *sawah* technology for rice production was introduced to these inland valleys in Nigeria. Sawah refers to a levelled and bounded rice field with inlet and outlet for irrigation and drainage (Wakatsuki et al., 1998). According to Wakatsuki and Masunaga (2005), *sawah* is a multifunctional constructed wetland which is a prerequisite for realization of the objectives of GR as well as maintaining a sustainable ecological environment. The geological fertilization

process and nitrogen fixation inherent in *sawah* based rice production system compensate for nutrients losses.

Sawah system was introduced in Nigeria in 1986 through on-farm adaptive research in the two research sites of Gara and Gadza inland valleys of Bida (Hirose and Wakatsuki, 2002). On these sites, the Japanese researchers conducted on-farm adaptive and participatory trials on Sawah for four years (1986–1990). In partnership with Watershed Initiative in Nigeria, a Non-Governmental Organization (NGO), Agricultural Development Project (ADP), Ministry of Agriculture, Niger state and National Cereals Research Institute (NCRI), the dissemination of the sawah technology took off in 2001 from villages previously identified in a diagnostic survey (Oladele and Wakatsuki, 2010). Since then, the dissemination and adoption have continued in other parts of Nigeria.

Sawah-based system of rice production was reported to have contributed to the achievement of GR in Asia. The speedy and remarkable scale with which the food problem was solved was unprecedented. This resulted in a substantial reduction in poverty and the beginning of broader economic growth in Asia. With GR, per capita production of rice has increased from 200 kg to more than 250 kg in the last 40 years in Asia. With proper soil management of the IVs, the yield of rice in Nigeria could improve to a level that can favourably compete with Asia thereby meeting the increasing demand for rice and contribute to food security in Nigeria.

As part of efforts to effectively utilize *sawah* technology in improving rice production level and to alleviate persistence shortage in rice supply, there is need to understand the physico-chemical and geochemical properties of

the sawah soils for the development and management of the inland valley ecosystems in Nigeria. Sustainable management of sawah in Nigeria will require a thorough evaluation of the soil fertility that is determined by physico-chemical and geochemical properties of the soils. Despite the importance of the physico-chemical and geochemical properties of the sawah soils in Nigeria, little information is available. Although Issaka et al., (1996), Buri et al., (2000) and Abe et al., (2007) conducted some basic soil surveys in inland valleys of West Africa, detailed study on the physico-chemical and geochemical properties of the sawah soils in Nigeria is required. This study, therefore, aims at investigating the physico-chemical and geochemical status of the soils under sawah in Nigeria. This will provide basic information for sustainable management of soils under sawah in Nigeria in order to meet up with the desired increase in rice production as well as rice demand and consumption among Nigerians. This study also provides recommendations for soil management for sustainable sawah rice production in Nigeria.

MATERIALS AND METHODS Study area and soil sampling

This study was carried out in five states in Nigeria where *sawah* rice production is being practiced. The states are Niger, Kaduna, Kwara, Ebonyi and Ondo. Data used in this study were collected in all the *sawah* sites in these states namely: Bida, Zaria,

Ilorin, Abakaliki, and Akure. The sites are Ejeti, Emir, Etusegi, Nasarafu, Shabamaliki, Sheshibikun, Etundandan, Zaria, Ilorin, Abakaliki (Ishiagu) and Akure as shown in Fig. 1. Soils of the Bida area are of Mesozoic (Cretaceous) origin, and are generally believed to be formed from Nupe sandstone. The soils of Zaria were derived from Basement Complex rocks which are essentially granites, gneisses, migmatites, schists and guartzites that are rich in guartz and low in divalent cations (Wall, 1978). The soils of Ilorin were formed from the Precambrian basement complex rocks and under the grassland savannah forest cover and belong to the soil group called ferruginous soil (Abe et al., 2007). The soils of Abakaliki were derived from Cretaceous black shale and siltstone or shale and limestone (Abe et al., 2007). The soils of Akure were made up of ferruginous tropical soils. Crystalline acid rocks constitute the main parent material of these soils. The main features include a sandy surface horizon underlain by a weakly developed clayey, mottled and occasionally concretionary sub-soil. The soil is however prone to erosion and occasional water logging as a result of the clay sub-soil. The soils have an exceptional clayey texture, but combine good drainage and aeration with good properties of moisture and nutrient retention. The climatic information of the study area is shown in Table 1. Soil samples were collected at depth of 0-15 cm, 15-30 cm, 30-45 cm and 45-60 cm using auger sampler from all the sites.



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Fig 1: Map of Nigeria showing the study locations

Site	State	Relative hu- midity (%)	Temperature (°C)	Annual Rainfall (mm)	Altitude (m)
EJT 1	Niger				97
EJT 2	Niger				92
EMR 1	Niger				76
EMR 2	Niger	45 - 87	23-34.	1100 -1200	82
ETS 1	Niger				77
ETS 2	Niger				73
NSF	Niger				71
SHB	Niger				91
SHE 1	Niger				71
SHE 2	Niger				76
ETD	Niger				126
ZNK 1	Kaduna				598
ZNK 2	Kaduna	20-85	10- 42	1000 -1200	593
ILA 1	Kwara				384
ILA 2	Kwara	75 - 80	34- 53	1130-1800	381
ILA 3	Kwara				376
ISH 1	Ebonyi				54
ISH 2	Ebonyi	60 – 80	20- 38	1500 -2000	47
AKR 1	Ondo				367
AKR 2	Ondo	80	28- 31	1405 - 2400	345

Table	1:	Sam	pling	locations
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Note: EJT = Ejeti; EMR = Emi; ETS = Etusegi; NSF = Nasarafu; SHB = Shabamaliki; SHE = Sheshibikun; ETD = Etundandan; ZNK = Zaria, ILA = Ilorin, ISH = Abakaliki (Ishiagu) and AKR = Akure.

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Laboratory analyses

Soil samples were air-dried, ground and passed through a 2 mm mesh sieve. Soil pH (H₂O and KCl) was measured with a pH meter (with a glass electrode) according to the method recommended by IITA (1979) and Mclean (1982) with a ratio of 1:2.5. Electrical Conductivity (EC) was determined using EC meter with glass electrode with a ratio of 1:5. Exchangeable cations were first extracted with 1M ammonium acetate solution pH 7. Exchangeable Ca (exch. Ca) and exchangeable Mg (exch. Mg) were determined using Inductive Coupled Plasma Atomic Emission Spectroscopy (Shimadzu ICPE 9000, Kyoto, Japan). Exchangeable K (exch. K) and exchangeable Na (exch. Na) were determined using atomic absorption spectrophotometer (AA-680; Shimadzu, Kyoto, Japan). Available phosphorus (avail. P) content was determined by Bray 2 method (Bray & Kurtz, 1945). Available Sulfur (avail. S) was first extracted with Di-potassium Hydrogen Phosphate (KH₂PO₄) and Inductively Coupled Plasma Atomic Emission Spectroscopy (Shimadzu ICPE 9000, Kyoto, Japan) was used to determine the avail. S. Available Silica (avail. SiO₂) was determined by colorimetric molybdenum blue method by extracting with acetate buffer with ascorbic acid as described by Yanai et al., (1996). Soil was extracted with 1M acetate buffer at pH 4.0 at a ratio of 1:10 for 5 hours at 40°C with occasional shaking. After filtration with dry filter paper no 6, the concentration of Si was determined using the colorimetric molybdenum blue method. Total Carbon (TC) and Total Nitrogen (TN) were determined by the dry combustion method using N-C analyzer (MT-700 J-Science Co. Ltd., Kyoto, Japan) based on the same principle described by Nelson and Sommers (1982). For

particle size distribution, sieving was employed to determine coarse sand (2.0-0.2 mm) and fine sand (0.2-0.02 mm), and the pipette method was used for silt (0.02-0.002 mm) and clay (<0.002 mm). Total elements were analysed using X-ray Fluorescence Spectrometry (XRF). The dried soil samples were ground for 20 minutes in an automatic agate mortar and pestle. Total SiO_2 , TiO_2 , AI_2O_3 , Fe_2O_3 (total iron expressed as Fe_2O_3), MnO, Na₂O, MgO, K₂O, CaO, and P₂O₅ abundance in the samples were determined by XRF in the Department of Geoscience, Shimane University, using a RIX-2000 spectrometer (Rigaku Denki Co. Ltd.) equipped with Rh-anode X-ray tube. All samples were made on pressed powder disks, following Ogasawara (1987). Powdered soil samples were ignited at 1050 °C prior to major element analyses. Loss on Ignition (LOI) of samples was determined before preparing glass beads. A 1.8 g sample of soil was mixed with 3.6 g flux powder (a mixture of Li2B4O7: LiBo2 at ratio of 4:1), placed in a platinum crucible and put in a bead sampler machine (NT-2000, Tokyo Kagaku, Co) for making glass beads. The glass beads of soil samples were then placed in an XRF spectrometer (RIX-2000 spectrometer, Rigaku Denki Co. Ltd.) to determine major elements content. Ten elements (Si, Fe, Al, Ti, Mn, Mq, Ca, Na, K, and P) were analysed.

Weathering degree

The degree of weathering was determined by calculating the Chemical Index Alteration (CIA) (Fedo *et al.*, 1995). The CIA reflects the proportion of primary and secondary minerals measured in soil samples and provides accurate measure of the degree of chemical weathering (Nesbitt and Young, 1982). This was calculated using the formula below:

 $CIA = AI_2O_3 / (AI_2O_3 + CaO + Na_2O + K_2O)$ (1)

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The mineral alteration is regarded as low when CIA is in range of 50 to 60, intermediate when it is between 60 and 80 and extreme for CIA greater than 80 (Fedo *et al.*, 1995).

Statistical analysis

Statistical analysis was performed to determine the relationships between the soil fertility characteristics. Correlation analysis was used to determine the relationship that existed among soil fertility properties using the Statistical Package for Social Science (SPSS version 16).

RESULTS

Topsoil physico-chemical and geochemical properties

Soil particle size of soil texture

The result of the study showed that *sawah* topsoils in Nigeria are majorly sandy loam to sandy clay loam in nature (Table 2) with average sand content of 60.60%. Majority of the sites had values above 50% except in ETS 2, AKR 2, EMR 1 and EJT 1 with 35%, 45%, 48% and 24% sand, respectively. The soils were low in both clay and silt with average values of 19.22% and 20.18%, respectively. Sand content across all the sampling sites ranged between 24.31% and 81.99%, silt content ranged between 9.18% and 52.94% while clay content ranged between 5.38% and 36.36%. As shown in Table 2, the topsoils are categorised as sandclay loam found in AKR 1, EMR 1, EMR 2 and ILA 3; sandy clay found in AKR 2, silty loam found in EJT 1, loamy sand found in EJT 2 and ZNK 2; clay loam found ETS 2 and sandy loam found in ETS 1, ILA 1, ILA 2, NSF, SHB 1, ZNK 1, ISH 1, ISH 2, SHE 1, SHE 2 and ETD. In most cases, sand is higher than clay and silt added together.

Chemical characteristics

Generally, the soils under *sawah* in Nigeria showed low topsoil pH values. Topsoil pH ranged between 4.6 and 6.8 with a mean value of 5.2 for in H_2O . Topsoil pH in KCl ranged between 3.7 and 5.8 with a mean value of 4.2. Both pH in H₂O and pH in KCI followed the same trend across the sampled soils. The pH was found to be moderately acidic to slightly acidic in the study area with few exceptions in Akure sites that showed slightly alkaline pH. Topsoil Electrical Conductivity (EC) values ranged between 0.007 dSm⁻¹ and 0.066 dSm⁻¹ with average of 0.016 dSm⁻¹. Although the values are low, the EC falls within the recommended limit of 0 - 4.0 dSm⁻¹ for crop production. As shown in Table 3, topsoil exch. Ca ranged between 0.57 cmol kg⁻¹ and 21.22 cmol kg⁻¹ with topsoil average of 3.51 cmol kg⁻¹. The values of exch. Ca were generally low across all the sampling locations. Highest value of exch. Ca was observed in Akure sites. Exch. K content of topsoil was low with mean content of 0.32 cmol kg-1. Exch. K showed a range of between 0.12 cmol kg⁻¹ and 0.60 cmol kg⁻¹ with topsoil average of 0.32 cmol kg⁻¹. Soils under *Sawah* in Nigeria are characterised by low exch. K with highest values observed in Akure. Exch. Mg was low. The topsoil exch. Mg ranged between 0.11 cmol kg⁻¹ and 5.63 cmol kg⁻¹ with topsoil average of 0.99 cmol kg⁻¹. Topsoil exch. Na ranged between 0.09 cmol kg⁻¹ and 1.49 cmol kg⁻¹ and with topsoil average of 0.37 cmol kg⁻¹. Ex. Na was generally low in soils under sawah in Nigeria. Topsoil Avail. P was low, ranging from 4.36 mg kg⁻¹ to 323.35 mg kg⁻¹ with topsoil average of 41.12 mg kg⁻¹. There was considerably high avail. P in AKR 1 and 2. AKR 1 and 2 are located in the Southwest part of Nigeria. This high value agreed with Issaka et al., (1996) that reported that soils in Southwest Nigeria had high Avail. P com-

pared to other inland valleys in West Africa. Topsoil avail. SiO₂ values ranged between 24.65 mg kg⁻¹ and 688.46 mg kg⁻¹ with topsoil average of 130.71 mg kg⁻¹. There was considerably high avail. SiO₂ in Akure which considerably affected the average value. Avail. S level was generally low across all the sampling locations compared to the critical level of 8 mg kg⁻¹ as recommended by Yamaguchi (1997). Topsoil avail. S values ranged between 3.56 mg kg⁻¹ and 31.25 mg kg⁻¹ with topsoil mean value of 9.57 mg kg⁻¹. Although the average value is higher than the critical level, however majority of the sites have values below the critical level. Highest value of avail. S was observed in Akure sites which considerably affected the average value. The content of TN was generally low in soils under sawah in Nigeria. The content of topsoil TN ranged between 0.24 g kg⁻¹ and 3.01 g kg⁻¹ with topsoil average value of 0.80 g kg⁻¹. Following a similar trend, topsoil TC can be categorised as low. The content of TC ranged between 2.94 g kg⁻¹ and 29.10 g kg⁻¹ with topsoil average value of 8.97 g kg⁻¹.

Geochemical properties and weathering degree

The result of the study as shown in Table 3 revealed that topsoil total elemental SiO_2 ranged between 67.57% and 94.47% in soils under *sawah*. The average topsoil SiO_2 is

85.96%. Majority of the locations sampled had values above the average. TiO₂ ranged between 0.60% and 6.39% with an average of 1.77%. Majority of the locations sampled had values above the average TiO_2 value. Al₂O₃ values ranged between 2.60 % and 13.63% with a mean of 7.61%. Fe_2O_3 ranged between 0.49% and 9.18% with an average of 2.59%. The average value of MnO was 0.06% and ranged between 0.02% and 0.27%. MgO ranged between 0.06% and 0.70% with an average of 0.18%. CaO values ranged between 0.08% and 1.83% with a mean of 0.33%. Na₂O ranged between 0.17% and 1.24% with an average of 0.42%. The result of the soils in Nigeria with average of 1.05%. P₂O₅ values on the other hand ranged between 0.01% and 0.23% with a mean of 0.04%. The result also showed that SiO₂ Al₂O₃ and Fe₂O₃ dominated, total elemental composition accounting for a cumulative average of 96.16%. Except TiO₂ and K_2O which showed average values >1%, MnO, MgO, CaO, Na₂O and P₂O₅ showed average values < 1%. The result of the study further revealed that soils under sawah have CIA values ranging from 67.45 and 91.50. Based on the interpretation of CIA value as prescribed by Fedo et al., (1995), soils under sawah in Nigeria exhibited intermediated (CIA 60 - 80) to extreme degree weathering (CIA>80).

extural Class		ind clay loam	indy clay	It loam	oamy sand	indy clay loam	indy clay loam	indy loam	lay loam	indy loam	indy loam	indy clay loam	indy loam	indy loam	indy loam	oamy sand	indy loam				
F	pui (%	23 Se	.05 Sc	.31 Si	.86 L	.75 Sč	46 Sc	.72 Sa	.28 C	45 Sc	.60 Se	.13 Se	.97 Se	.72 Sã	.44 Se	-99 L	.84 Se	.42 Se	:09 Se	62 Se	:09 Sã
	e Sa	8 63	1 45	0 24	3 79	8 47	8 60	0 57	7 35	4 68	6 54	9 55	5 75	3 64	5 77	2 81	7 67	7 55	2 73	3 63	5 60
	F.Sar (%)	27.2	32.8	23.0	53.7	43.5	44.5	52.0	34.1	27.4	29.0	30.6	37.8	31.5	52.5	62.1	55.6	52.2	50.0	49.9	56.8
	C.Sand (%)	35.95	12.24	1.31	26.14	4.17	15.88	5.72	1.11	41.01	25.54	24.44	38.13	33.20	24.89	19.87	12.17	3.15	23.08	13.69	3.24
	Silt (%)	12.88	20.31	52.94	14.75	29.45	13.95	27.30	28.36	14.64	24.43	15.37	10.95	19.13	9.18	11.98	13.21	24.42	19.40	16.98	23.95
	Clay (%)	23.89	34.64	22.75	5.38	22.79	25.60	14.99	36.36	16.91	20.97	29.50	13.07	16.15	13.37	6.03	18.95	20.16	7.51	19.40	15.96
	C/N	11.06	9.67	11.99	13.06	12.08	10.18	11.67	10.96	11.28	10.52	11.76	16.04	12.28	12.63	13.21	11.03	11.00	11.09	10.14	11.28
	gkg₁	15.04	29.10	17.86	4.70	5.47	4.38	6.55	6.25	9.81	3.05	14.94	3.85	7.37	7.20	9.38	10.37	10.23	3.66	2.94	7.22
	g kg [.]	1.36	3.01	1.49	0.36	0.45	0.43	0.56	0.57	0.87	0.29	1.27	0.24	0.60	0.57	0.71	0.94	0.93	0.33	0.29	0.64
	/ail S g kg-i)	8.67	1.25	0.88	5.07	7.18	5.84	.95	1.78	9.45	6.85	9.87	3.56	5.71	6.33	0.44	3.39	5.75	5.15	5.24	3.06
	₹Ē	2	ç	-	2,		C	÷		0		0		c	÷	-		÷		C	~
	Avail SiC (mg kg ⁻¹	135.60	688.46	106.39	81.11	107.57	145.10	89.43	130.11	260.30	174.44	24.65	28.25	63.42	53.24	39.73	55.01	77.98	54.62	74.36	224.34
	Bray-2 P (mg kg ^{.i})	90.21	323.35	49.21	21.87	8.30	12.96	12.77	10.79	29.26	4.36	54.95	6.95	32.59	15.85	19.03	34.54	20.58	5.46	6.06	63.25
ol kg	Na	0.75	1.49	0.22	0.37	0.40	0.42	0.29	0.26	0.49	0.41	0.16	0.28	0.16	0.56	0.09	0.27	0.14	0.22	0.16	0.29
ions (cm	Mg	2.74	5.63	0.77	0.37	0.54	0.63	0.41	0.54	1.12	0.79	0.62	1.19	0.89	1.73	0.13	0.35	0.11	0.43	0.31	0.54
geable cat 1)	¥	0.40	0.48	0.49	0.12	0.48	0.36	0.31	0.36	0.37	0.58	0.27	0.13	0.19	09.0	0.13	0.50	0.16	0.15	0.13	0.21
Exchan	Ca	7.11	21.22	3.56	1.72	1.77	2.09	2.29	2.75	3.54	2.74	1.47	2.34	1.83	5.79	0.58	1.37	0.57	2.32	1.50	3.58
	EC (d Sm ^{.1})	0.03	0.07	0.02	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.02	0.01	0.02	0.02	0.01	0.01	0.01	0.01
_	KCI	4.34	5.75	4.28	4.76	3.95	3.74	4.23	3.92	3.79	4.02	3.91	4.55	4.65	4.24	4.42	4.12	3.73	4.03	4.05	3.96
Hd	H ₂ 0	5.13	6.43	4.93	5.42	4.85	4.75	4.98	4.86	4.64	4.79	4.63	6.11	4.98	6.82	5.08	4.76	4.71	5.25	5.97	4.75
	Sites	AKR 1	AKR 2	EJT 1	EJT 2	EMR 1	EMR 2	ETS 1	ETS 2	ILA 1	ILA 2	ILA 3	NSF 1	SHB 1	ZNK 1	ZNK 2	ISH 1	ISH 2	SHE 1	SHE 2	ETD 1
	s/N	-	2	e	4	5	9	7	8	6	10	11	12	13	14	15	16	17	18	19	20

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Site	SiO ₂ (%)	TiO ₂ (%)	AI ₂ O ₃ (%)	Fe²O₃ (%)	MnO (%)	MgO (%)	CaO (%)	Na ₂ O (%)	K2O (%)	P ₂ O5 (%)	ΓΟΙ	CIA
AKR 1	73.77	6.39	8.96	8.03	0.10	0.45	1.09	0.58	0.56	0.07	4.49	80.07
AKR 2	67.57	5.04	13.61	9.18	0.27	0.70	1.83	0.62	0.93	0.23	9.26	80.09
EJT 1	85.77	2.29	8.87	1.48	0.03	0.13	0.19	0.21	0.99	0.05	5.56	86.45
EJT 2	94.26	1.25	2.60	1.05	0.04	0.07	0.12	0.18	0.42	0.01	1.63	78.40
EMR 1	86.98	1.49	7.72	1.74	0.04	0.12	0.25	0.33	1.31	0.02	3.19	80.39
EMR 2	87.07	1.02	7.85	2.58	0.03	0.12	0.21	0.26	0.84	0.02	3.47	85.66
ETS 1	89.49	1.48	6.40	0.96	0.04	0.11	0.15	0.21	1.14	0.02	3.14	81.13
ETS 2	80.24	1.73	13.63	2.07	0.04	0.17	0.20	0.26	1.63	0.03	5.13	86.71
ILA 1	88.36	1.54	6.10	3.13	0.05	0.13	0.17	0.22	0.29	0.04	4.02	90.03
ILA 2	84.49	2.18	7.87	4.21	0.07	0.19	0.20	0.23	0.51	0.04	3.28	89.26
ILA 3	88.65	1.62	6.91	1.85	0.04	0.14	0.20	0.24	0.32	0.03	4.66	90.09
NSF 1	94.47	0.97	2.89	0.49	0.02	0.06	0.09	0.19	0.81	0.01	1.24	72.58
SHB 1	91.32	1.25	5.34	0.90	0.02	0.08	0.08	0.18	0.82	0.02	2.51	83.10
ZNK 1	82.05	09.0	10.38	1.77	0.03	0.16	0.34	0.71	3.96	0.01	2.53	67.45
ZNK 2	85.51	0.66	8.37	1.37	0.03	0.14	0.42	0.73	2.76	0.01	2.64	68.18
ISH 1	88.25	0.72	6.58	2.34	0.03	0.21	0.22	1.22	0.40	0.04	2.84	78.24
ISH 2	84.38	0.89	9.11	3.15	0.02	0.28	0.22	1.24	0.67	0.04	3.35	81.00
SHE 1	90.37	1.10	4.96	1.56	0.18	0.08	0.26	0.31	1.17	0.01	1.59	74.08
SHE 2	88.18	1.17	7.32	1.41	0.03	0.11	0.29	0.31	1.18	0.01	2.52	80.53
ETD 1	88.03	1.92	6.74	2.50	0.04	0.11	0.10	0.17	0.36	0.04	3.38	91.50
CIA = C	hemical Ind	lex Alter	ation; LO	I = Loss on Ig	nition							

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Ca K Mg		3 0.56	5 0.81 0.47	9 0.69 0.63 0.64	5 0.12 0.00 -0.01 -0.	2 0.38 0.28 0.12 0.	2 0.02 0.24 -0.17 -0.	9 0.01 0.23 -0.05 -0.	3 0.02 0.20 -0.01 -0.	4 0.05 -0.27 0.36 -0.	9 0.03 0.40 0.01 -0.	0.12 0.28 -0.17 -0.	9 0.09 -0.40 0.11 0.	4 0.29 -0.55 -0.05 -0.	1 0.25 0.28 0.02 0.	1 0.24 0.47 0.01 0.	3 0.10 0.46 -0.03 0.	9 0.04 -0.10 -0.14 0.	3 0.15 0.36 -0.22 -0.	2 0.06 0.17 -0.13 -0.	5 0.29 0.10 -0.27 -0.	5 0.38 0.22 0.33 0.	0.16 0.44 -0.09 -0.	1 0.05 0.17 -0.09 0.
Va D5 Sid					0.21	0.49 0.19	0.24 0.63 0	0.28 0.74 -0	0.29 0.74 -0	0.03 0.04 -0	0.01 0.14 0	0.16 0.23 0	0.11 0.23 -0	0.12 0.07 -0	0.06 0.34 0	0.04 0	0.32 0.04 0	0.00 0.25 0	0.07 0.03 0	0.08 0.30 -0	0.19 0.03 -0	0.16 0.33 -0	0.02 0.54 0	0.05 0.47 0
02 S							.20	.01 0.82	.08 0.82	1.44 -0.24	.21 0.30	.23 0.35	-0.40	.21 -0.40	.55 0.27	.14 0.42	.63 0.18	.03 -0.25	.06 0.26	.30 0.30	.32 0.13	.34 0.03	.53 0.55	.68 0.33
z									0.99	-0.11	0.33	0.44	-0.47	-0.25	0.25	0.27	0.11	-0.27	0.38	0.04	0.26	-0.14	0.65	0.32
υ										0.01	0.26	0.43	-0.42	-0.20	0.23	0.23	0.02	-0.30	0.31	0.06	0.23	-0.08	0.58	0.24
C/N											-0.45	-0.22	0.38	0.46	-0.27	-0.44	-0.63	-0.26	-0.42	-0.16	-0.11	0.22	-0.46	-0.53
Clay %												0.37	-0.78	-0.53	0.44	0.63	0.35	-0.25	0.39	-0.11	-0.08	-0.20	0.47	0.63
Silt													-0.87	-0.28	0.73	0.33	0.06	0.03	0.12	-0.17	-0.22	-0.17	0.59	0.42
Sand														0.47	-0.73	-0.56	-0.22	0.12	-0.29	0.17	0.19	0.22	-0.64	-0.62
siO ₂															-0.12	-0.97	-0.52	0.12	-0.68	-0.55	-0.34	-0.50	-0.46	-0.10
TIO ₂																0.14	0.29	0.11	-0.08	-0.43	-0.61	-0.45	09.0	0.79
Al ₂ O ₃																	0.37	-0.19	09.0	0.48	0.27	0.48	0.41	0.13
Fe ₂ O ₃ *																		0.10	0.67	0.09	0.26	-0.25	0.67	0.51
OnM																			-0.23	0.11	-0.20	-0.09	-0.12	-0.05
OgM																				0.31	0.76	0.04	0.62	0.12
CaO																					0.47	0.72	0.17	0.53
a ₂ D K ₂																					-	2 00 5	-0- -0- -).4 1 -0.
0																							88	0.0

Correlation analysis between soil fertility parameters

Correlation analysis shown in Table 4 revealed that there is a significant relationship between TC and Avail. S and Avail. P. Total Carbon was also found to be significantly related to total elemental P_2O_5 . The results also showed similar significant relationship between TN and Avail. S and Avail. P. Total Nitrogen was also found to be significantly related to total elemental P_2O_5 . Clay content showed a significant relationship with total elemental P₂O₅. Sand content showed a negative correlation TC, TN and total elemental P2O5. The result also revealed that CIA is correlated with Avail. P., Avail. SiO₂, clay content, total elemental CaO, Na₂O, K₂O, TiO₂, Fe₂O₃ and P₂O₅.

Distribution of soil fertility parameters with depth

Particle size analysis revealed that clay content of soils under sawah increased with increase in depth from topsoil down to subsoil except in few cases with erratic distribution. Silt content showed a decreasing trend in depth from topsoil down to subsoil. Silt content however showed variations in some sampling points without clear-cut trends. Sand showed erratic trend in most of the sampling points with variations in depth from topsoil down to subsoil. Soil pH increased with increase in depth from topsoil down to subsoil. The profile distribution revealed that the pH level increased with depth with few exceptions where the pH level decreased with depth. AKR 1, EMR 2, ETU 2, ILA 1, SHB, ZNK 2, ISH 1, ISH 2 and SHE 1 showed the same trend with an increase in the pH from topsoil down to subsoil. The EC values generally followed the opposite trend as pH with decrease in the value of EC with the depth from topsoil down to subsoil with only few exceptions.

With exception in few cases, exch. Ca decreased with increase in soil depth from topsoil to subsoil. Exch. K decreased with increase in depth from topsoil down to subsoil. Exch Mg increased slightly in some locations and also decreased with depth from topsoil down to subsoil in other locations. Exch. Na increased with depth from topsoil down to subsoil which may be due to leaching of the topsoil and accumulation in the subsoil. Avail. P decreased with depth from topsoil down to subsoil. Avail. SiO₂ values increased with depth from topsoil down to subsoil except in few occasions. Avail. S level was comparatively higher in the topsoil across all the sampling locations. Avail. S decreased with increase in depth from topsoil down to subsoil with some exceptions where there was erratic distribution within the soil profile. TN values were comparatively higher in the topsoil across all the sampling locations and decreased with increase in depth of the profile. As observed in TN, TC values were comparatively higher in the topsoil across all the sampling locations and decreased with increase in depth from topsoil down to subsoil.

DISCUSSION

The physical characteristics of a soil, particularly texture significantly influence the other characteristics of the soil. Particle size distribution is critical in relation to soil behaviour and management. The properties of individual particles and their distribution in the soils are subjected to limited human control. According to Lund *et al.*, (1999), the diverse ratios of sand, silt, and clay in the soil results in soil variations which has a direct effect on crop yield. These variances affect the waterholding capacity, nutrient leaching, and plant root stability in soils. Soil physical properties also play a significant role in the chemical properties of soil. As reported by previous 1C. I. ALARIMA, 1M. A. BUSARI, 1J. M. AWOTUNDE, 2O. O. OLANIYI, 3T. MASUNAGA AND 3T. WAKATSUKI

authors (Buri et al., 2000 and Issaka et al., 1996) inland valleys in West Africa are dominated by low clay content having been derived from granites and Pre-Cambrian metamorphic rocks, generally referred to as Basement Complex. The result revealed a positive correlation although not significant between sand and total elemental SiO_2 (r = 0.47) which means high sand content in soil increases the SiO₂ content. The high content of sand fraction means that the soils water retention capacity was low and as such could not hold nutrients resulting into high level of leaching which have contributed to the generally low fertility status of these soils. The low content of clay in these soils contributes to high leaching especially as observed in exch. Na which increases in values with increase in soil depth. The high content of clay in ETS 2 and silt in EJT 1 will be an advantage for these locations for sawah rice development. The clay content of soil play a major role in its nutrient supplying ability as well as it water holding capacity (Buri et al., 1999).

The pH was found to be moderately acidic to slightly acidic in the entire site but with few exceptions in Akure site that showed slightly alkaline pH. The acidic nature of parent rocks and intense leaching under high rainfall is responsible for the acidic reaction of the soils (Abe et al., 2007). Application of nitrogen fertilizers commonly adopted by farmers in Nigeria account for the acidic condition which in turn worsens the fertility of the soils. Electrical conductivity (EC) values ranged between 0.005 dSm⁻¹ and 0.26 dSm⁻¹ with topsoil average of 0.02 dSm⁻¹. The EC values were generally low and followed opposite trend as pH, decreasing in value with depth with only few exceptions. Chabra et al., (1980) and Pereir et al., (1986) reported that soils exposed to

high rainfall have low soluble salts concentration because of leaching losses and in turn results in decreased electrical conductivity. The EC values are low in the study area but fall within the recommended limit for crop production. Corresponding with the low pH, the contents of exchangeable bases were very low. The values of exch. Ca were generally low across all the sampling locations in Nigeria. Exch. K content of the soils was low with mean content of 0.26 cmol kg⁻¹. Exch. Na was generally low except in few occasions where exch. Na was moderate and increased slightly with depth of the soil profile. The low level of exchangeable bases is due to leaching of the topsoil and accumulation in the subsoil. In addition, flooding condition of the soils which reduced the pH at the surface also account for the nature of the exchangeable cations found in soils under sawah in Nigeria. Low exchangeable bases could also be associated with the low level of clay across all the sample locations. The soils are low in clay and hence may be susceptible to leaching. Buri et al., (2000) also reported that low colloidal activity as a result of low organic matter and erratic rainfall distribution and low clay activity in West Africa provided an environment where cations retention and overall build-up soil plant nutrient is very low. The low content of exchangeable bases is also due to high degree of weathering as shown by high CIA values obtained in this study. As prescribed by Fedo et al., (1995) soils under *sawah* in Nigeria exhibited intermediate CIA (60-80) to extreme degree of weathering (CIA>80). Majority of the soil sampled fell into the category of extreme degree of weathering. With extreme degree of weathering, rapid loss of mobile species such as base cations (Ca, Mg, K and Na) from soil is eminent which account for the results observed in this study. The high level of SiO2, Al₂O₃ and Fe₂O₃ and low level of

total base elements (K₂O, CaO, Na₂O and MgO) is also an indication of low fertility status obtained in this study. Avail. P in the soils under *sawah* in Nigeria is low. The result is supported by previous reports of Annan-Afful *et al.*, (2004) and Abe *et al.*, (2007) who reported an average avail. P of 4.9 mg kg⁻¹ in soils under *sawah* in Ashanti region of Ghana and a range of between 1.0 mg kg⁻¹ and 9.0 mg kg⁻¹ in eastern Nigeria, respectively.

As there has not been any study on availability of silica in lowland soils in Nigeria to the best of our knowledge, we compare the result to the critical level recommended by authors in tropical Asia and Japan. According to Sumida (1992), the critical value of avail. SiO₂ content for rice growth is 300 mg kg⁻¹ of SiO₂. Also, Bollich & Matichenkov (2002) described values less than 300 mg kg⁻¹ of SiO₂ as deficient and values less than 600 mg kg⁻¹ of avail. SiO₂ as low for rice and sugarcane. According to IRRI (2000), the critical value of avail. SiO₂ content for rice growth is 86 mg kg⁻¹. Based on Sumida (1992), soils under *sawah* in Nigeria are deficient in avail. SiO₂ except in Akure 2 where the topsoil recorded a high level of silica, other sites ranged from low to deficient in available silica. Based on IRRI (2000), majority of the soils had values below the critical level of 86 mg kg⁻¹. Silicon is as beneficial element for rice plants and as one of the major factors affecting the sustainability of rice production (Husnain *et al.*, 2008), sawah soils in Nigeria need silicate amendment for optimum rice production. Silicon can control rice diseases such as blast, sheath blight in rice, and powdery mildew in cucumber (Ishizuka & Hayakawa, 1951; Kawashima, 1927; Miyake & Takahashi 1983; Ma et al., 2001). Silica according to Iler (1979) is able to displace

phosphate ions from the soil surface. Ma *et al.*, (2001) also reported that silicon is essential in alleviating water stress by decreasing transpiration and is beneficial to rice under P deficiency (as found in the present study locations) and excess of P, Na, Mn, N and Al.

Avail. S level was low in the topsoil across all the sampling locations showing values below the critical level of 8 mg kg⁻¹ as recommended by Yamaguchi (1997). According to Yamaguchi (1997), sulfur status is a key factor controlling rice productivity. The result of this study is in agreement with previous results of Osiname and Kang (1975), Enwezor (1976) Kang et al., (1981) and Buri et al., (2000) who reported similar low level of sulfur in West Africa. The result also showed that avail. S significantly correlated with TC and TN. According to Buri et al., (2000), poor organic matter management has a great effect on sulfur deficiency. In addition to organic matter, volcanic activity which supply sulfur to soil through precipitation is absent in the study area (Buri et al., 2000) and may also account for the low level of sulfur observed in this study. Total Carbon and Total Nitrogen were observed to be low across all the soils under sawah. Organic matter according to Oyediran (1990) is effective in increasing and retaining most cations. Poor organic matter content resulting in low TC and TN plays a major role in low cation level found in this study. In addition, farming practices that encourage the burning of plant residue and annual burning of vegetation especially during the dry season (slash and burn) commonly found in the study areas also account for the low content of TC and TN. The values of TC and TN decreased with increase in depth down the soil profile. The comparably high TC and TN on the topsoil was as a result of large biomass

production in the tropical condition found in the study locations.

Total elemental SiO₂ Al₂O₃ and Fe₂O₃ dominated total elements, accounting for a cumulative average of 96.16%. Total elemental SiO₂ was the most abundant element accounting for between 67.57% and 94.47% of total weight. There also existed a positive correlation between total elemental SiO₂ and sand. This implies a strong chemical weathering process of the soils and sand size particles are dominated by SiO_2 . AI_2O_3 values ranged between 2.60 % and 13.63% with a mean of 7.61%. Fe₂O₃ ranged between 0.49% and 9.18% with an average of 2.59%. Both Al₂O₃ and Fe₂O₃ were positively correlated with clay contents (r = 0.63; r = 0.35, respectively) suggesting that Al₂O₃ occurs in the form of clay and alluminosilicate minerals, and consists of SiO₂ and Al₂O₃ produced through weathering of parent materials. The correlation also suggests that the clay fraction not only consists of alluminosilicate minerals but also free iron oxides, which is formed from iron released from parent materials during weathering processes. Total elemental MgO content was low and ranged between 0.06% and 0.70%. The low level of total elemental MgO may also be responsible for low level of Ex. Mg found in this study. P₂O₅ values ranged between 0.01% and 0.23% and positively correlated with clay content. The result was also supported by Yichu et al., (1984) who reported a similar relationship between phosphorus and clay. According to Lair et al., (2009), phosphorus retention and release was correlated with clay-sized particles. Clay-sized particles represent the soil fraction with the highest and most reactive surface area.

Base on the result of the study, Akure sites (AKR 1 and AKR 2) had peculiar characteristics among the locations. Akure sites were higher than other location in all the soil fertility parameters under investigation and hence affected the overall result. The result from Akure may be as a result of few reasons which include geological fertilization, fertilizer application and vegetation. Akure is located in the southwest of Nigeria with thick forest. According to Issaka et al., (1996), in the equatorial forest zone, baserich sediments of Tertiary-Quaternary, relating to volcanic activities may account for the presence of soils rich in exchangeable cations, phosphorus, high pH and other fertility parameters found in Akure sites. Further research is intended to ascertain the reason (s) for this variation.

The general low level of soil fertility parameters found in the study areas is due to the high level of weathering. As reported in Table 2, the CIA values of the soils revealed that sawah soils in Nigeria fall within intermediated (CIA 60-80) to extreme weathering rate (CIA>80) which may result in high level of leaching. Extreme degree of weathering leads to rapid loss of mobile species such as base cations. The role of climate cannot also be over emphasised. Climate seems to exert a great effect on the availability of organic matter which has a pronounced effect on the amount of TN, TC, avail. P, exchangeable cations and avail. S. High temperature and rainfall of the study area as shown in Table 1 could aid the rapid decomposition of organic matter. High precipitation coupled with the sandy nature of soils could also lead to high rate of leaching of major nutrients especially exchangeable bases.

CONCLUSION

This study investigated the physicochemical and geochemical compositions of the soils under *sawah* in Nigeria. This study has provided useful information needed to improve sawah rice production thereby increase rice productivity in Nigeria. The study revealed that soils under *sawah* in Nigeria are majorly sandy loam to sandy clay loam. The study also indicated that soils under sawah are acidic and lacking basic fertility parameters such as exchangeable bases, TC, TN, avail. S avail. P and avail. SiO₂. The result further revealed that total elemental SiO₂ Al₂O₃ and Fe₂O₃ dominated total elements in soils under *sawah* Nigeria. Soils under Sawah exhibited intermediated to extreme weathering rate with majority of the soil sampled falling into the category of extreme weathering rate. With extreme degree of weathering, rapid loss of mobile species such as base cations (Ca, Mg, K and Na) from soil is imminent resulting in low fertility status of soils under sawah. Leaching of major nutrients due to high sand proportion in the soil, high rainfall and low level of organic matter also contributed to the low fertility status of the soils.

As soil texture cannot be easily influenced by management practices, to ameliorate the trend of depletion in the fertility, combination of conservative agricultural practices which encourage organic matter accumulation and recycling is recommended. Therefore, effective organic matter management is necessary and a key factor in improving the basic cations of these soils. Poor farm management through removal of crop biomass (such as rice straw) by farmers used for feeding livestock and other purposes should be discouraged while encouraging recycling crop biomass with the soil to increase soil organic matter content. Adop-

tion of leguminous crops in rotation with rice especially after harvest of rice is also recommended as this will increase nitrogen through bacteria fixation and reduce the use of nitrogen fertilizers thereby reducing soil acidification process. Due to the importance of silica in rice production, a detail study which will provide a recommended rate of silica towards improving rice production in Nigeria is needed. Also, a study to ascertain the dynamics of nutrient in lowland in Akure site is recommended.

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